

## Origin and temperature of formation of ubiquitous carbonate veins in low grade metamorphic rocks and hydrothermal ore deposits: a clumped isotope perspective

### Contact persons

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### Context

In the realm of low-temperature diagenetic but also low-grade regional and contact metamorphic settings, neotectonism, as well as ore deposit research, the use of  $\Delta_{47}$ -measurements provides exciting new research avenues to help understand the petrologic evolution. For example, clumped isotope compositions measured for marbles, carbonatites as well as limestones and dolomites that have also been influenced by diagenesis and low-grade metamorphism (Ferry et al., 2011), apparent temperatures of crystallization recorded by clumped isotope compositions were between 175 and 300 °C. As peak metamorphic temperatures for marbles are clearly higher, the temperatures obtained may represent final blocking temperatures to internal re-ordering or solid-state diffusion. The converse of this is that the clumped isotope thermometry can provide a new set of constraints on the extents and mechanisms of post-depositional modifications of carbonate. In contrast, low temperature metamorphic vein carbonate or carbonate in hydrothermal veins may well preserve their clumped isotope composition and hence give true temperatures of crystallization, which in turn constrains the fluid-rock interactions for such mineralization. These possibilities are to be evaluated in the context of both regional and contact metamorphic domains within the Alps, as well as for hydrothermal vein carbonates that are so ubiquitous in ore deposits.

### Aims and Methods

An exciting new development in stable isotope geochemistry over the past decade is the measurement of "isotopic clumping" in molecular CO<sub>2</sub> and in crystallized carbonate from which the CO<sub>2</sub> is extracted by acid digestion. This "clumping", which refers to the tendency of having the heavy but rare stable isotopes of carbon and oxygen occurring together in the same molecule of CO<sub>2</sub> relative to their stochastic abundance within the same phase (expressed as  $\Delta_{47}$ ), is thermodynamically favored (Eiler and Schauble, 2004). The measurement of isotopic clumping thus has the advantage over conventional measurements of isotopic compositions of CO<sub>2</sub> in that it provides a single-mineral, thermodynamically based geothermometer. Given knowledge on the temperature of carbonate crystallization in the often mono-mineralogic veins, the isotopic composition of fluids in equilibrium with the carbonate can then also be determined, which further constrains the fluid-rock interactions during vein formation.

### References

Eiler, J.M., 2007. "Clumped Isotope" geochemistry. *The study of naturally occurring, multiply-substituted isotopologues. Earth and Planet. Sci. Lett.* 262, 309-327.

Ferry, J.M., Passey, B.H., Vasconcelos, C., Eiler, J.M., 2011. Formation of dolomite at 40 to 80 °C in the Latemar carbonate buildup, Dolomites, Italy, from clumped isotope thermometry. *Geology* 39, 571 - 574.

1 image dimension H5.5cm x L8.5cm



### Website

[www.unil.ch/idyst/en/home/menuinst/lab-facilities/stable-isotope-laboratory.html](http://www.unil.ch/idyst/en/home/menuinst/lab-facilities/stable-isotope-laboratory.html)

### Prerequisite

Indicate if the student must take some course or module